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Synthesis of the alkenyl-substituted tetracyclic core of the bisabosquals

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Abstract—HCl-catalyzed deprotection and cyclization of benzylic alcohol 15 cleanly provided tricycle 16 by a cis-selective intramolecular Diels–Alder reaction. Acetylation of the phenol, bis epoxidation, and base-catalyzed hydrolysis and cyclization afforded tetracycle 19 with the bisabosqual skeleton, but the opposite stereochemistry at the tertiary alcohol stereocenter. Selective dehydration of the tertiary alcohol to form the exocyclic alkene, ozonolysis, reductive deoxygenation of the side chain epoxide, and addition of MeMgBr to the ketone from the less hindered face gave tertiary alcohol 24 with the tetracyclic core of bisabosqual A (1). © 2007 Elsevier Ltd. All rights reserved.

1. Introduction

Bisabosqual A (1) was isolated in 2001 from the culture broth of Stachybotrys sp. RF-7260, obtained from decaying tree leaves.[1](#page-5-0) Three related natural products, bisabosquals B–D, were isolated from Stachybotrys ruwenzoriensis RF-6853. Bisabosqual A (1) has broad spectrum antifungal activity in vitro and inhibits the microsomal squalene synthases from Saccharomyces cerevisiae, Candida albicans, HepG2 cell, and rat liver with IC_{50} values of 0.43, 0.25, 0.95, and 2.5 µg/mL, respectively, suggesting that bisabosqual A might be useful for the treatment of hypercholesterolemia.

Scheme 1. Retrosynthetic analysis of bisabosqual A.

The novel tetracyclic structure of bisabosqual A (1) was determined by 2D NMR experiments and confirmed by X-ray crystallography of bisabosqual B.[2](#page-6-0) The three six-membered rings of 1 are analogous to those of tetrahydrocannabinoids (THCs), although the cyclohexane and pyran rings are transfused in the extensively investigated $THCs³$ $THCs³$ $THCs³$ and cis-fused in 1. The additional furan ring and tertiary alcohol of 1 pose additional synthetic challenges.

We envisaged that bisabosqual $A(1)$ might be accessible by oxidative cyclization of cis-fused tricycle 2, which should be available by an inverse electron demand Diels–Alder reaction of quinone methide 3 (see Scheme 1). Although hexahydrocannabinoids are invariably formed with a trans ring fusion, Rickards found that treatment of 4 with TMSCl and Et_4 NBr cleaved the MOM ethers and generated a quinone methide that cyclized to give 65% of the cis-fused tetrahydrocannabinoid tricycle 5 (see Eq. 1).[4](#page-6-0)

These observations are consistent with MMX calculations of transition state energies for the intramolecular Diels–Alder reaction.^{[5](#page-6-0)} The transition state leading to a *trans*-fused hexahydrocannabinoid is more stable than the lowest energy

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transition state leading to a cis-fused hexahydrocannabinoid by 1.2 kcal/mol. The presence of the double bond in the tether changes the transition state energies such that the one leading to the cis-fused tetrahydrocannabinoid 5 is more stable than the one leading to the *trans*-fused tetrahydrocannabinoid by 0.6 kcal/mol.

2. Results and discussion

Deprotonation of the bis MOM ether of orcinol $(6)^6$ $(6)^6$ $(6)^6$ at the 2position with n-BuLi in THF followed by the addition of citral (7) afforded 90–100% of 8 as a \approx 60:40 *E/Z* mixture that was used without purification (see Scheme 2).^{[7,8](#page-6-0)} In our hands, treatment of crude alcohol 8 with TMSBr,^{[9](#page-6-0)} TMSCl and Et₄NBr, or TMSCl and Bu₄NBr in CH₂Cl₂ at 25 °C did not give tricyclic phenol 9. Use of TMSBr at -78 °C provided the MOM ether of 9 (not shown) in 60% yield, which can also be obtained in 70% yield with TMSCl and NaI at -20 °C.^{[10](#page-6-0)} Eventually, we found that complete cyclization of 8 and deprotection to give 9 can be effected in 58% overall yield from 6 by heating 8 in a 1:6 mixture of 3 M aqueous hydrochloric acid and MeOH at 60° C for 2 h. Tricycle 9 is somewhat unstable and can only be purified on MeOHdeactivated silica gel.

Scheme 2. Synthesis of tetracyclic model 14.

syn Oxidative cyclization^{[11,12](#page-6-0)} of 9 would yield the desired tertiary alcohol 14 in a single step. However, all previous oxidative cyclizations have started with unsaturated alcohols, rather than phenols. The electron rich resorcinol is more easily oxidized than the alkene of 9. Treatment of 9 with bis(collidine)iodonium hexafluorophosphate 13 afforded the diiodo phenol. Reaction with $Hg(OAc)$ ₂ also occurred on the aromatic ring.^{[14](#page-6-0)} Not surprisingly, treatment of 9 with $(CF_3CO_2)ReO_3$, $(Cl_2HCCO_2)ReO_3$, or PCC led to complex mixtures of products.

Epoxidation of 9 with *m*-CPBA in CH_2Cl_2 takes place at the alkene from the less hindered α -face as reported by Razdan in a similar system.^{[15](#page-6-0)} The initially formed epoxy phenol 10 partially cyclized to give tetracycle 11 with the desired ring system, but the opposite stereochemistry at the tertiary alcohol stereocenter. Treatment of this mixture with methanolic NaOH for 2 h provided 83% of 11.

The rigidity of the ring system allowed us to develop an efficient procedure to convert 11 to the desired alcohol 14. Treatment of 11 with MsCl and excess Et_3N in CH_2Cl_2 provided 89% of an 84:16 mixture of 12 and the endocyclic isomer. Formation of the less stable alkene 12 is favored because only the methyl protons can adopt the required antiperiplanar orientation to the equatorial leaving group. Oxidative cleavage of the alkene mixture with $OsO₄$ and $NaIO₄$ gave ketone 13. Addition of MeMgBr to the ketone in THF occurred selectively from the less hindered α -face to afford the desired tertiary alcohol 14 in 64% yield from alkene 12. The spectral data of the cyclohexanol protons and carbons of 14 correspond closely to those of bisabosqual A (1), while those of 11 are quite different. In particular, Me-COH absorbs at δ 1.25 in 14 and δ 1.31 in 1, but at δ 0.87 in the epimeric alcohol 11, in which the methyl group of the tertiary alcohol is in the shielding cone of the aromatic ring.

We then turned our attention to the preparation of 24, containing the unsaturated side chain of the bisabosquals. Deprotonation of 6^6 6^6 with *n*-BuLi in THF followed by addition of $6E$ -farnesal^{[16](#page-6-0)} afforded 15 as an E/Z mixture that partially decomposed on chromatography (see [Scheme 3\)](#page-2-0). Although benzylic alcohol 15 could be isolated in 73% yield, the overall yield of acetate 17 from 6 was higher when crude 15 was used for the intramolecular Diels–Alder reaction. Cyclization of 15 and deprotection to give phenol 16 were achieved by heating 15 in a 1:6 mixture of 3 M aqueous hydrochloric acid and MeOH in a 60 °C oil bath for 3 h. The yield of 16 decreased at higher temperatures. Chromatographic purification was best carried out after acetylation of crude phenol 16 in 1:1 Ac₂O/pyridine for 12 h at 25 °C to give acetate 17. This three-step sequence afforded 90% pure 17 in 48% overall yield from 6.

We were initially disappointed to find that epoxidation of either 16 or 17 with 1 equiv of m-CPBA occurred selectively on the side chain double bond rather than on the desired cyclohexene double bond. However, on further consideration, epoxidation of the side chain double bond would protect this double bond during the oxidative cleavage of the exocyclic double bond of 20 that generates cyclohexanone 22.

We therefore epoxidized 17 with 2.5 equiv of *m*-CPBA in CH₂Cl₂ at 25 °C for 1 h to give bis epoxide 18 as a 1:1 mixture of diastereomers on the side chain epoxide. Hydrolysis of the acetate and cyclization with K_2CO_3 in MeOH at 25 °C for 45 min afforded tetracyclic alcohol 19 in 54% overall

Scheme 3. Synthesis of alkenyl-substituted tetracyclic model 24.

yield from dienyl acetate 17. The spectra of the two diastereomers of 19 were surprisingly different, suggesting that they might differ in the stereochemistry at one or more of the ring positions, rather than simply at the epoxide on the side chain. Fortunately, Cornforth reductive deoxygenation 17 of the diastereomeric mixture with Zn, NaI, and NaOAc in HOAc afforded a single compound establishing that only a mixture of side chain epoxides was present in 19.

Conversion of tertiary alcohol 19 to the exocyclic alkene 20 and then to ketone 22 proved much more challenging than for model 11 without the functionalized side chain. Treatment of 19 with MsCl and excess Et_3N in CH_2Cl_2 provided 4:1 to 8:1 mixtures of 20 and 21 in only 5–30% yield. Dehydration of 19 with Martin's sulfurane,^{[18](#page-6-0)} Ph₂S(OC) $(CF_3)_2Ph$ ₂, in CH_2Cl_2 proceeded cleanly, but with considerable loss of regioselectivity, to give an inseparable 2:1 mixture of 20 and 21. Oxidative cleavage of the double bond of 20 with $OsO₄/NaIO₄$ with or without 2,6-lutidine^{[19](#page-6-0)} proceeded in low yield. Ozonolysis of the 2:1 mixture of 20 and 21 in CH_2Cl_2 containing pyridine at -78 °C followed by reduction with activated $\overline{\text{Zn}}^{20}$ $\overline{\text{Zn}}^{20}$ $\overline{\text{Zn}}^{20}$ provided epoxy ketone 22 in 43% overall yield from epoxy alcohol 19. Reductive deoxygenation of 22 by Cornforth's procedure^{[17](#page-6-0)} with Zn(Cu), NaI, and NaOAc in HOAc at 25° C for 30 min afforded alkenyl ketone 23 in 53% yield. Slightly lower yields were obtained with activated Zn or Zn(Ag) and much lower yields were obtained with unactivated Zn dust. Epoxy ketone 23 was obtained in 0–20% yield with $\text{Cp}_2\text{TiCl}_2/\text{Zn}$, 21 21 21 WCl₆/ BuLi, 22 22 22 or NaI/TMSCl in CH₃CN.^{[23](#page-6-0)}

Addition of MeMgBr to ketone 23 in THF occurred selectively from the less hindered α -face to afford the desired tertiary alcohol 24 in 78% yield. The spectral data of the cyclohexanol protons and carbons of 24 correspond closely to those of bisabosqual $A(1)$, while those of 19 are quite different. In particular, MeCOH absorbs at δ 1.24 in 24 and at δ 1.31 in 1, but at δ 0.86 and 0.87 in the two diastereomers of the epimeric epoxy alcohol 19, in which the methyl group of the tertiary alcohol is in the shielding cone of the aromatic ring (see Fig. 1). The stereochemistry at C_7 of 24 was established by an NOE from H₅ at δ 3.61 to H₈ at δ 1.73–1.64 and δ 1.61–1.52, but not to H₁₄ at δ 1.39.

The spectra of 1 and 24 are very similar except for H_4 , C_4 , and C_7 (see [Table 1](#page-3-0) and Fig. 1 for atom numbering scheme). The absorptions are further downfield in both the ${}^{1}H$ and ${}^{13}C$ NMR spectra of 1 as expected, because the two aldehydes of 1 are electron withdrawing whereas the aromatic methyl group of 24 is electron donating. The methoxy protons of 25, which are analogous to H_4 of 1, absorb at δ 3.93 in $CDCl₃,²⁴$ $CDCl₃,²⁴$ $CDCl₃,²⁴$ whereas the methoxy protons of 26, which are analogous to H₄ of 24, absorb at δ 3.76 in CCl₄.^{[25](#page-6-0)}

In conclusion, we have developed a short and efficient route to the tetracyclic core of the bisabosquals that effectively deals with the side chain unsaturation. We are currently

Figure 1. Three-dimensional structures of 19 and 24.

Table 1. Comparison of the ${}^{1}H$ and ${}^{13}C$ NMR spectra of bisabosqual A (1) and tetracyclic model 24

	['] H NMR		$^{13}\mathrm{C}$ NMR	
	1	24	1	24
1	1.55, 1.28	$1.8 - 1.4$, $1.3 - 1.1$	16.3	16.4
\overline{c}	1.79, 1.21	$1.8 - 1.4$, $1.3 - 1.1$	34.9	35.1
3			69.1	69.5
4	4.97 (d, 8.8)	4.69 (d, 8.5)	93.8	90.5
5	3.66 (dd, 8.8, 6.6)	3.61 (dd, $8.5, 6.1$)	33.3	33.7
6	2.05	1.96-1.88	35.9	36.5
7			83.5	81.0
8	1.67, 1.57	$1.8 - 1.4$	38.7	38.3
9	2.08	$2.12 - 2.03$	22.2	22.4
10	5.03	5.04	123.1	123.7
11			132.5	131.9
12	1.65	1.65	25.6	25.6
13	1.59	1.59	17.6	17.6
14	1.46	1.39	22.1	22.2
15	1.31	1.24	29.5	29.8

adapting this to more highly functionalized resorcinols needed for the synthesis of the bisabosquals.

3. Experimental section

3.1. General procedures

NMR spectra were recorded at 400 MHz in CDCl₃ unless otherwise indicated. Chemical shifts are reported in δ , coupling constants in hertz, and IR spectra in cm^{-1} .

3.2. a-(2,6-Dimethyl-1,5-heptadienyl)-2,6-bis(methoxymethoxy)-4-methyl-benzenemethanol (8)

n-BuLi (6.6 mL of a 1.6 M solution in hexanes, 10.6 mmol) was added to a solution of 6^{6b} 6^{6b} 6^{6b} (2 g, 9.43 mmol) in THF (88 mL) at 0° C. The reaction mixture was warmed to 25 °C and stirred for 2 h. The reaction mixture was cooled to 0° C and citral (7) (2.1 mL, 12.1 mmol) in THF (15 mL) was added dropwise. The reaction mixture was then stirred at 25° C for 3 h, quenched with NH₄Cl, and extracted with Et₂O (3×40 mL). The combined extracts were washed with brine, dried $(MgSO₄)$, and concentrated under reduced pressure to yield 3.78 g (110%) of 8 as a \approx 6:4 E/Z mixture that was about 90% pure by 1 H NMR analysis: 1 H NMR 6.63 (s, 2), 5.900 (dd, 0.6×1 , J=10.9, 9.1), 5.897 (dd, 0.4×1 , $J=12.2, 9.1$, 5.70 (br d, 0.4 \times 1, J=9.1), 5.68 (br d, 0.6 \times 1, $J=9.1$), 5.24–5.18 (m, 4), 5.06 (t, 1, $J=6.7$), 3.59 (d, $0.6\times1, J=10.9, OH$, 3.53 (d, $0.4\times1, J=12.2, OH$), 3.50 $(s, 0.6 \times 6), 3.49$ $(s, 0.4 \times 6), 2.30$ $(s, 3), 2.08 - 2.20$ $(m, 2),$ 2.08–1.95 (m, 2), 1.64 (s, 3), 1.55 (s, 3).

3.3. (6aR,10aS)-rel-6a,7,8,10a-Tetrahydro-3,6,6,9-tetramethyl-6H-dibenzo $[b,d]$ pyran-1-ol (9)

A 3 M solution of HCl (22 mL, 65 mmol) was added to a solution of crude 8 (2.0 g, 5.49 mmol) in MeOH (130 mL). The reaction mixture was heated in a 60° C oil bath for 2 h, cooled to 25 °C, quenched with saturated NaHCO₃, and extracted with $Et₂O$. The combined extracts were washed with brine, dried $(MgSO₄)$, and concentrated under reduced pressure to give 1.48 g (104%) of crude 9. Flash chromatography of the residue on MeOH-deactivated silica gel (hexanes) gave 825 mg (58%) of tricycle 9 that was 90–95% pure. Further chromatography on non-deactivated silica gel resulted in decomposition to give less pure 9: ¹H NMR 6.24 (s, 1), 6.22 (br s, 1), 6.12 (s, 1), 4.92 (br s, OH), 3.58–3.53 (br, 1), 2.18 (s, 3), 2.00–1.89 (m, 3), 1.74– 1.66 (m, 1), 1.68 (s, 3), 1.52–1.42 (m, 1), 1.39 (s, 3), 1.27 (s, 3); 13C NMR 154.8, 153.8, 137.3, 135.0, 121.8, 110.7, 109.3, 108.7, 76.2, 40.0, 31.4, 29.7, 25.9, 25.2, 23.7, 20.9, 20.6; HRMS (DEI) calcd for $C_{17}H_{22}O_2$ (M⁺) 258.1620, found 258.1612.

3.4. (3R,3aR,9aR,9bS)-rel-2,3,3a,9,9a,9b-Hexahydro-3,6,9,9-tetramethyl-1H-benzofuro[4,3,2-cde][1]benzopyran-3-ol (11)

A solution of m-CPBA (276 mg, 1.357 mmol) in CH_2Cl_2 (20 mL) was added dropwise to a solution of 9 (254 mg, 0.984 mmol) in CH_2Cl_2 (54 mL) at 0 °C. The reaction mixture was warmed to 25° C and stirred for 12 h. The solvent was evaporated and the bright orange residue was dissolved in Et₂O. The solution was then washed with $Na₂SO₃$, NaHCO₃, and brine, dried (MgSO₄), and concentrated under reduced pressure. The residue was taken up in MeOH (40 mL) and 4% NaOH (25.1 mL) was added to the solution. The reaction mixture was stirred at 25 °C for 2 h. The MeOH was evaporated and the aqueous phase was extracted with $Et₂O$. The combined extracts were washed with brine, dried $(MgSO₄)$, and concentrated under reduced pressure. Flash chromatography of the residue on MeOH-deactivated silica gel (hexanes) gave 224 mg (83%) of 11: ¹H NMR 6.19 (s, 1), 6.16 (s, 1), 4.85 (br d, 1, $J=8.6$), 3.72 (dd, 1, $J=8.6, 6.7$), 2.26 $(s, 3)$, 1.97 (ddd, 1, J=11.6, 6.7, 6.1), 1.74–1.65 (m, 2), 1.48–1.36 (m, 1), 1.41 (s, 3), 1.34 (s, 3), 0.96 (dddd, 1, $J=14.3$, 11.0, 11.0, 4.3), 0.87 (s, 3); ¹³C NMR 161.2, 151.9, 140.4, 107.57, 107.51, 102.4, 93.6, 79.0, 73.2, 37.6, 35.2, 34.8, 26.7, 26.0, 24.6, 22.1, 19.3; HRMS (DEI) calcd for $C_{17}H_{22}O_3$ (M⁺) 274.1569, found 274.1558.

3.5. (3aR,9aR,9bS)-rel-2,3,3a,9,9a,9b-Hexahydro-6,9,9-trimethyl-3-methylene-1H-benzofuro[4,3,2-cde]- [1]benzopyran (12)

MsCl (0.8 mL, 10.2 mmol) was added dropwise to a solution of 11 (157 mg, 0.573 mmol) and Et_3N (2.7 mL, 18.8 mmol) in CH_2Cl_2 (23 mL) at 0 °C. The reaction mixture was warmed to 25° C and stirred for 14 h. The reaction was then quenched with 2 M HCl and extracted with $Et₂O$. The combined extracts were washed with brine, dried (MgSO4), and concentrated under reduced pressure. Flash chromatography of the residue on MeOH-deactivated silica gel (hexanes) gave 130 mg (89%) of an 84:16 mixture of 12 and the endocyclic isomer: 1 H NMR (12) 6.25 (s, 1), 6.16 (s, 1), 5.33 (br d, 1, $J=7.9$), 5.14 (br s, 1), 4.81 (br s, 1), 3.68 (br dd, 1, $J=7.9, 7$), 2.27 (s, 3), 2.24 (ddd, 1, $J=12, 3, 3$), 2.00 (ddd, 1, $J=12, 6, 6$), 1.88–1.75 (m, 2), 1.40 (s, 3), 1.34 (s, 3), $0.92 - 0.82$ (m, 1); ¹H NMR (partial data for endocyclic isomer) 5.50 (br d, 1, $J=5.2$), 5.20 (br d, 1, $J=8$), 3.77– 3.72 (m, 1), 2.25 (s, 3); 13C NMR 160.4, 151.9, 144.8, 140.0, 110.3, 107.7, 106.7, 103.3, 86.6, 78.7, 39.7, 36.5, 31.8, 26.4, 26.2, 23.7, 22.2; HRMS (DEI) calcd for $C_{17}H_{20}O_2$ (M⁺) 256.1463, found 256.1471.

3.6. (3aR,9aR,9bS)-rel-1,2,3a,9,9a,9b-Hexahydro-6,9,9 trimethyl-3H-benzofuro[4,3,2-cde][1]benzopyran-3-one (13)

OsO₄ (42 μ L of a 2.5% solution in *t*-BuOH, 0.004 mmol) and NaIO_4 (53 mg, 0.246 mmol) were added to a solution of $12(21 \text{ mg}, 0.082 \text{ mmol})$ in THF/H₂O $(2:1, 1 \text{ mL})$. The reaction mixture was stirred at 25° C for 48 h and concentrated under reduced pressure. The residue was taken up in $H₂O$ and extracted with EtOAc. The combined extracts were washed with $Na₂S₂O₃$ and brine, and dried (MgSO₄). Flash chromatography of the residue on MeOH-deactivated silica gel (9:1 hexanes/EtOAc) gave 19 mg (92%) of 80% pure ketone 13: ¹H NMR 6.36 (s, 1), 6.18 (s, 1), 5.04 (d, $1, J=7.3$, 4.07 (dd, 1, $J=7.3, 7.3$), 2.40–2.28 (m, 3), 2.26 (s, 3), 2.10–2.05 (m, 1), 1.46 (s, 3), 1.40 (s, 3), 1.38–1.23 (m, 1); ¹³C NMR 208.4, 160.5, 151.3, 141.1, 108.3, 104.6, 103.3, 87.4, 78.5, 39.1, 39.0, 38.7, 26.7, 26.4, 22.7, 22.1.

3.7. (3S,3aR,9aR,9bS)-rel-2,3,3a,9,9a,9b-Hexahydro-3,6,9,9-tetramethyl-1H-benzofuro[4,3,2-cde][1]benzopyran-3-ol (14)

MeMgBr (0.21 mL, 0.214 mmol) was added to a solution of partially purified ketone 13 (14 mg, 0.054 mmol) in THF (1 mL) at 0 °C. The reaction mixture was warmed to 25 °C and stirred for 1 h, quenched with $NH₄Cl$, and extracted with EtOAc. The combined extracts were washed with brine, dried (MgSO4), and concentrated under reduced pressure. Flash chromatography of the residue on MeOH-deactivated silica gel (hexanes) gave 10.4 mg (70%) of 14 : ¹H NMR 6.22 $(s, 1), 6.17 (s, 1), 4.70 (br d, 1, J=8.5), 3.66 (br d, 1, J=8.5,$ 6), 2.26 (s, 3), 1.84 (ddd, 1, $J=11.6$, 6, 6), 1.71 (br d, 1, $J=11.6$), 1.60–1.40 (m, 1), 1.42 (s, 3), 1.33 (s, 3), 1.25 (s, 3), 1.26–1.16 (m, 2); 13C NMR 161.4, 151.6, 140.3, 108.1, 107.6, 101.4, 90.5, 78.8, 69.4, 38.6, 34.9, 34.1, 29.6, 26.7, 26.1, 22.2, 16.5; HRMS (DEI) calcd for $C_{17}H_{22}O_3$ (M⁺) 274.1569, found 274.1570.

3.8. a-(2,6,10-Trimethyl-1,5,9-undecatrienyl)-2,6-bis- (methoxymethoxy)-4-methyl-benzenemethanol (15)

n-BuLi (7.0 mL, 1.6 M in hexane, 11.2 mmol) was added at 0° C to a solution of 6 (1.70 g, 8.0 mmol) in THF (50 mL). The resulting solution was warmed to 25° C and stirred for 4 h. A THF solution (12 mL) of a 2:1 mixture of $(2E, 6E)$ and $(2Z,6E)$ -farnesal^{[16](#page-6-0)} (2.46 g, 11.2 mmol) was added dropwise to the reaction mixture and the reaction mixture was stirred at 25 \degree C for 4 h. The reaction was quenched with saturated NH₄Cl (40 mL) and extracted with $Et₂O$ (3×50 mL). The combined $Et₂O$ extracts were dried (MgSO₄) and concentrated to give crude 15 (3.40 g). A portion of the crude product (43 mg) was purified by flash chromatography on MeOH-deactivated silica gel (15:1 hexanes/EtOAc) to yield 15 (23 mg) as a mixture of cis and trans isomers: ${}^{1}H$ NMR 6.62 (s, 2), 5.91 (dd, 1, $J=9.2$, 9.2), 5.68 (d, 1, $J=9.2$), 5.21 (s, 2), 5.20 (s, 2), 5.11–5.04 (m, 2), 3.68–3.60 (m, 1, OH), 3.49 (s, 6), 2.29 (s, 3), 2.24–2.21 (m, 1), 2.11–1.89 (m, 7), 1.80 (s, 3), 1.67 (s, 3), 1.58 (br s, 3), 1.55 (br s, 3); 13C NMR 154.8 (2C), 138.7, 137.3, 135.1, 131.2, 126.6, 124.3, 123.8, 118.6, 109.0 (2C), 94.4 (2C), 64.3, 56.2 (2C), 39.64, 39.60, 26.6, 26.3, 25.6, 21.8, 17.6, 16.4, 15.9; IR (neat) 3318, 2919, 2854, 1664, 1611; HRMS (EI⁺) calcd for $C_{26}H_{40}O_5$ (M⁺) 432.2876, found 432.2871.

3.9. (6S,6aR,10aS)-rel-6a,7,8,10a-Tetrahydro-3,6,9-trimethyl-6-(4-methyl-3-pentenyl)-6H-dibenzo[b,d]pyran-1-yl acetate (17)

Aqueous hydrochloric acid (30 mL, 3 M) was added dropwise to a solution of crude 15 (3.38 g) in MeOH (180 mL) at 60 °C. The resulting solution was heated in a 60 °C oil bath for 3 h and cooled to 25 °C. The reaction was quenched with saturated NaHCO₃ (40 mL) and extracted with $Et₂O$ $(3\times50 \text{ mL})$. The combined Et₂O extracts were washed with brine, dried $(MgSO₄)$, and concentrated to give crude 16.

Acetic anhydride (6 mL) was added to a pyridine (6 mL) solution of crude 16. The reaction mixture was stirred at 25 \degree C for 12 h. The resulting mixture was diluted with $Et₂O$ (60 mL) and washed with H_2O (2×30 mL), NaHCO₃ $(3\times30 \text{ mL})$, and brine (30 mL), dried (MgSO₄), and concentrated to give crude 17. Flash chromatography on MeOHdeactivated silica gel (40:1 hexanes/EtOAc) yielded 1.41 g (48% for three steps) of 90% pure 17 as a colorless oil: $\mathrm{^{1}H}$ NMR 6.51 (s, 1), 6.40 (s, 1), 5.86–5.81 (br, 1), 5.07–5.00 (br, 1), 3.45–3.39 (br, 1), 2.33 (s, 3), 2.22 (s, 3), 2.06–1.86 (m, 3), 1.81–1.41 (m, 6), 1.67 (s, 3), 1.63 (s, 3), 1.53 (s, 3), 1.37 (s, 3); 13C NMR 169.0, 153.4, 149.8, 137.2, 135.3, 131.7, 123.9, 121.1, 115.9, 115.3, 115.2, 78.2, 37.4, 37.0, 31.2, 29.6, 25.7, 23.7, 22.9, 22.3, 21.4, 20.9, 20.2, 17.4; IR (neat) 2966, 2927, 1766; HRMS (Q-TOF) calcd for $C_{24}H_{33}O_3$ (MH⁺) 369.2430, found 369.2437.

3.10. (1aR,3aR,4S,9bS,9cS)-rel-4-[(RS)-2-(3,3-Dimethyloxiranyl)ethyl]-1a,2,3a,4,9b,9c-hexahydro-1a,4,7-trimethyl-3H-oxireno[3,4]benzo[1,2-c][1]benzopyran-9-yl acetate (18)

A solution of m-CPBA (1.05 g, 70%, 4.3 mmol) in CH_2Cl_2 (30 mL) was added to a solution of 17 (630 mg, 1.7 mmol) in CH_2Cl_2 (12 mL). The reaction mixture was stirred for 1 h at 25 °C and concentrated. The residue was redissolved in EtOAc (60 mL), washed with saturated $Na₂SO₃$ $(2\times30 \text{ mL})$, NaHCO₃ ($2\times30 \text{ mL}$), and brine (30 mL), dried $(MgSO₄)$, and concentrated to give 558 mg of crude 18 as an oil, which was used directly in the next step. A portion of the crude product (28 mg) was purified by flash chromatography on MeOH-deactivated silica gel (20:1 hexanes/ EtOAc) to yield bis epoxide 18 (10 mg) as a 1:1 mixture of diastereomers: ¹H NMR 6.53 (s, 1), 6.51 (s, 1), [3.16– 3.14 (br, 1), 3.13–3.11 (br, 1)], [3.11 (d, 1, $J=6.1$), 3.08 (d, 1, J=6.1)], $[2.75$ (t, 1, J=6.1), 2.74 (t, 1, J=6.1)], $[2.37 (s, 3), 2.36 (s, 3)], 2.27 (s, 3), 2.03-1.03 (m, 9),$ 1.31 (s, 3), [1.28 (s, 3), 1.26 (s, 3)], 1.26 (s, 3), 1.24 (s, 3); 13C NMR (169.48, 169.43), (154.21, 154.17), (149.57, 149.54), 138.37, (115.79, 115.76), (115.10, 115.05), (112.81, 112.80), (77.65, 77.49), (64.27, 64.17), (62.27, 62.24), (58.41, 58.37), (58.14, 58.09), (35.47, 35.14), 34.00, 32.34, (25.97, 25.88), 24.80, (23.10, 23.04), (22.75, 22.56), 21.93, (21.05, 21.03), 20.96, (19.38, 19.28), (18.63, 18.44); IR (neat) 2958, 2927, 1768; HRMS (Q-TOF) calcd for $C_{24}H_{33}O_5$ (MH⁺) 401.2328, found 401.2327.

3.11. (R,3aR,9S,9aR,9bS)-rel-9-[(RS)-2-(3,3-Dimethyloxiranyl)ethyl]-2,3,3a,9,9a,9b-hexahydro-3,6,9-trimethyl-1H-benzofuro[4,3,2-cde][1]benzopyran-3-ol (19)

 K_2CO_3 (1.0 g) was added to a solution of crude 18 (548 mg) in MeOH (15 mL). The reaction mixture was stirred at 25 $\mathrm{^{\circ}C}$ for 45 min. The mixture was diluted with saturated $NH₄Cl$ (15 mL) and extracted with EtOAc $(3 \times 30 \text{ mL})$. The combined EtOAc extracts were dried $(MgSO₄)$ and concentrated to give crude 19. Flash chromatography on MeOH-deactivated silica gel (3:2 hexanes/EtOAc) yielded 330 mg (54% from 17) of 19 as a 1:1 mixture of diastereomers: 1 H NMR 6.19 (s, 1), [6.15 (s, 1), 6.14 (s, 1)], [4.87 (d, 1, $J=8.0$), 4.85 (d, 1, $J=8.0$)], [3.71 (dd, 1, $J=8.0$, 7.4), 3.66 (dd, 1, $J=8.0, 7.4$], [2.69 (t, 1, $J=6.1$), 2.65 (t, 1, $J=6.1$)], 2.26 (s, 3), 2.08–2.00 (m, 2), 1.93–1.53 (m, 6), 1.52–1.29 (m, 1), [1.38 (s, 3), 1.35 (s, 3)], 1.27 (s, 3), 1.25 (s, 3), 1.05– 0.93 (m, 1), $[0.87 \text{ (s, 3)}, 0.86 \text{ (s, 3)}];$ ¹³C NMR (161.23, 161.17), (151.64, 151.60), (140.48, 140.42), (107.57, 107.54), (107.44, 107.42), (102.58, 102.53), (93.74, 93.63), (80.71, 80.67), (73.15, 73.11), (64.23, 63.88), (58.49, 58.31), (35.98, 35.57), (34.97, 34.92), (34.88, 34.82), (34.79, 34.71), (24.77, 24.75), (24.48, 24.40), (23.63, 23.43), 22.37, (22.15, 22.11), (19.22, 19.16), (18.61, 18.57); IR (neat) 3432, 2946, 2870, 1623; HRMS $(Q$ -TOF) calcd for $C_{22}H_{31}O_4$ (MH⁺) 359.2222, found 359.2221.

3.12. (3aR,9S,9aR,9bS)-rel-9-[(RS)-2-(3,3-Dimethyloxiranyl)ethyl]-1,2,3a,9,9a,9b-hexahydro-6,9-dimethyl-3H-benzofuro[4,3,2-cde][1]benzopyran-3-one (22)

A solution of Martin's sulfurane (562 mg, 0.84 mmol) in dry CH_2Cl_2 (10 mL) was added to a solution of 19 (200 mg, 0.56 mmol) in dry CH_2Cl_2 (5 mL) at 0 °C. The resulting solution was warmed to 25 $\mathrm{^{\circ}C}$ and stirred for 3 h. The reaction mixture was concentrated to give a 2:1 mixture of 20 and 21.

The residue was dissolved in CH_2Cl_2 (25 mL) and pyridine (0.3 mL). The mixture was cooled to -78 °C. Ozone was bubbled through it for 12 min while the reaction was monitored by TLC (every 60 s). The ozone flow was replaced by an air flow and the reaction was quenched with the addition of Zn (300 mg, activated) at -78 °C. The mixture was slowly warmed to 25 °C over 1 h and stirred at 25 °C for an additional 2 h. The resulting mixture was filtered. The filtrate was concentrated to give crude 22. Flash chromatography on MeOH-deactivated silica gel (1:1 hexanes/ EtOAc) yielded 82 mg (43% from 4) of 22 as a 1:1 mixture of diastereomers: ¹H NMR 6.36 (s, 1), [6.18 (s, 1), 6.17 (s, 1)], [5.06 (d, 1, J=7.8), 5.04 (d, 1, J=7.8)], [4.07 (dd, 1, $J=7.8, 7.2$, 4.02 (dd, 1, $J=7.8, 7.2$), [2.71 (t, 1, $J=6.0$), 2.67 (t, 1, J=6.0)], 2.42–2.28 (m, 3), 2.25 (s, 3), 2.12–2.04 (m, 1), 1.97–1.57 (m, 5), [1.43 (s, 3), 1.40 (s, 3)], 1.29 (s, 3), 1.27 (s, 3); 13C NMR (208.25, 208.17), (160.56, 160.50), 151.10, (141.24, 141.18), (108.42, 108.30), (104.63, 104.51), (103.54, 103.51), (87.47, 87.43), 80.29, (64.11, 63.71), (58.55, 58.35), (38.78, 38.76), 38.66, (37.34, 37.05), (35.03, 34.81), (24.79, 24.77), (23.65, 23.48), (22.77, 22.71), (22.66, 22.50), 22.15, (18.67, 18.64); IR (neat) 2966, 2928, 1732, 1624; HRMS (Q-TOF) calcd for $C_{21}H_{27}O_4$ (MH⁺) 343.1909, found 343.1917.

3.13. (3aR,9S,9aR,9bS)-rel-9-(4-Methyl-3-pentenyl)- 1,2,3a,9,9a,9b-hexahydro-6,9-dimethyl-3H-benzofuro $[4,3,2\text{-}cde][1]$ benzopyran-3-one (23)

A mixture of sodium acetate (18 mg, 0.22 mmol), sodium iodide (62 mg, 0.41 mmol), and zinc–copper couple (54 mg, 0.83 mmol) was added to a solution of 22 (71 mg, 0.21 mmol) in acetic acid (0.6 mL). The resulting solution was stirred at 25° C for 30 min. The reaction was quenched with saturated NaHCO₃ (10 mL) and extracted with EtOAc $(3\times15 \text{ mL})$. The combined EtOAc extracts were dried $(MgSO₄)$, and concentrated to give crude 23. Flash chromatography on MeOH-deactivated silica gel (5:1 hexanes/ EtOAc) yielded 36 mg $(53%)$ of pure 23 as a colorless oil: ¹H NMR 6.35 (s, 1), 6.18 (s, 1), 5.06 (t, 1, J=7.2), 5.04 (d, $1, J=7.6$, 4.03 (dd, 1, $J=7.6$, 6.4), 2.42–2.31 (m, 3), 2.25 (s, 3), 2.18–2.02 (m, 3), 1.77–1.53 (m, 2), 1.67 (s, 3), 1.61 (s, 3), 1.43 (s, 3), 1.34–1.22 (m, 1); ¹³C NMR 208.3, 160.5, 151.3, 141.1, 132.2, 123.4, 108.4, 104.7, 103.3, 87.5, 80.8, 38.9, 38.7, 38.4, 36.9, 25.6, 22.8, 22.7, 22.4, 22.1, 17.7; IR (neat) 2966, 2917, 1730, 1623; HRMS (Q-TOF) calcd for $C_{21}H_{27}O_3$ (MH⁺) 327.1960, found 327.1965.

3.14. (3S,3aR,9S,9aR,9bS)-rel-9-(4-Methyl-3-pentenyl)- 2,3,3a,9,9a,9b-hexahydro-3,6,9-trimethyl-1H-benzofuro[4,3,2-cde][1]benzopyran-3-ol (24)

MeMgBr (0.24 mL, 1.4 M in toluene/THF, 0.33 mmol) was added to a solution of 23 (27 mg, 0.083 mmol) in THF (10 mL) at $0 °C$. The resulting solution was warmed to 25° C and stirred for 5 h. The reaction was quenched with saturated NH4Cl (10 mL) and extracted with EtOAc $(3\times15 \text{ mL})$. The combined EtOAc extracts were dried $(MgSO₄)$ and concentrated to give crude 24. Flash chromatography on MeOH-deactivated silica gel (15:1 hexanes/ EtOAc) yielded 22 mg $(78%)$ of pure 24 as a colorless oil: ¹H NMR 6.20 (s, 1), 6.17 (s, 1), 5.04 (t, 1, J=7.6, H₁₀), 4.69 (d, 1, J=8.5, H₄), 3.61 (dd, 1, J=8.5, 6.1, H₅), 2.25 $(s, 3)$, 2.12–2.03 (m, 2, 2H₉), 1.96–1.88 (m, 1, H₆), 1.75– 1.45 (m, 4, H₁, H₂, and 2H₈), 1.65 (s, 3), 1.59 (s, 3), 1.39 $(s, 3, H_{14}), 1.33-1.13$ (m, 2, H₁ and H₂), 1.24 (s, 3, H₁₅); ¹³C NMR 161.4, 151.6, 140.3, 131.9, 123.7, 108.1, 107.7, 101.4, 90.5, 81.0, 69.5, 38.3, 36.5, 35.1, 33.7, 29.8, 25.6, 22.5, 22.4, 22.2, 17.6, 16.4; IR (neat) 3442, 2965, 1716, 1626; HRMS (EI⁺) calcd for $C_{22}H_{30}O_3$ (M⁺) 342.2195, found 342.2201. The relative stereochemistry of the side chain of 24 was established by a 1D NOESY experiment. Irradiation of H₅ at δ 3.61 showed an NOE to H₈ at δ 1.73–1.64 and δ 1.61–1.52, but not to H₁₄ at δ 1.39.

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References and notes

1. Minagawa, K.; Kouzuki, S.; Nomura, K.; Yamaguchi, T.; Kawamura, Y.; Matsushima, K.; Tani, H.; Ishii, K.; Tanimoto, T.; Kamigauchi, T. J. Antibiot. 2001, 54, 890–895.

- 2. Minagawa, K.; Kouzuki, S.; Nomura, K.; Kawamura, Y.; Tani, H.; Terui, Y.; Nakai, H.; Kamigauchi, T. J. Antibiot. 2001, 54, 896–903.
- 3. (a) Mechoulam, R.; Ben-Shabat, S. Nat. Prod. Rep. 1999, 16, 131–143; (b) Razdan, R. K. The Total Synthesis of Natural Products; ApSimon, J., Ed.; Wiley and Sons: New York, NY, 1981; Vol. 4, pp 185–262.
- 4. (a) Moore, M.; Rickards, R. W.; Rønneberg, H. Aust. J. Chem. 1984, 37, 2339–2348; For related observations, see: (b) Cruz-Almanza, R.; Pérez-Flores, F.; Lemini, C. Heterocycles 1994, 37, 759–774; (c) Tapia, R. A.; Alegría, L.; Valderrama, J. A.; Cortés, M.; Pautet, F.; Fillion, H. Tetrahedron Lett. 2001, 42, 887–889.
- 5. PCMODEL version 8.0 from Serena Software was used with **MMX**
- 6. (a) Townsend, C. A.; Davis, S. G.; Christensen, S. B.; Link, J. C.; Lewis, C. P. J. Am. Chem. Soc. 1981, 103, 6885–6888; (b) Ohta, S.; Nozaki, A.; Ohashi, N.; Matsukawa, M.; Okamoto, M. Chem. Pharm. Bull. 1988, 36, 2239–2243.
- 7. For a preliminary report on the preparation of 14, see: Snider, B. B.; Lobera, M. Tetrahedron Lett. 2004, 45, 5015–5018.
- 8. For another approach to the bisabosquals, see: Zhou, Z.; Parker, K. A. Abstracts of Papers, 228th National Meeting of the American Chemical Society, Philadelphia, PA, Aug 22–26, 2004; American Chemical Society: Washington, DC, 2004; ORGN 760.
- 9. Hanessian, S.; Delorme, D.; Dufresne, Y. Tetrahedron Lett. 1984, 25, 2515–2518.
- 10. Rigby, J. H.; Wilson, J. Z. Tetrahedron Lett. 1984, 25, 1429– 1432.
- 11. (a) Towne, T. B.; McDonald, F. E. J. Am. Chem. Soc. 1997, 119, 6022–6028; (b) McDonald, F. E.; Schultz, C. C. Tetrahedron 1997, 53, 16435-16448; (c) González, I. C.; Forsyth, C. J. J. Am. Chem. Soc. 2000, 122, 9099–9108; (d) McDonald, F. E.; Singhi, A. D. Tetrahedron Lett. 1997, 38, 7683–7686.
- 12. Schlecht, M. F.; Kim, H.-j. J. Org. Chem. 1989, 54, 583–587.
- 13. Homsi, F.; Robin, S.; Rousseau, G. Org. Synth. 2000, 77, 206– 211.
- 14. Sandin, R. B. J. Am. Chem. Soc. 1929, 51, 479–483.
- 15. Uliss, D. B.; Razdan, R. K.; Dalzell, H. C.; Handrick, G. R. Tetrahedron 1977, 33, 2055–2059.
- 16. Xiao, X.-y.; Prestwich, G. D. Synth. Commun. 1990, 20, 3125– 3130.
- 17. Cornforth, J. W.; Cornforth, R. H.; Mathew, K. K. J. Chem. Soc. 1959, 112–127.
- 18. Martin, J. C.; Arhart, R. J. J. Am. Chem. Soc. 1971, 93, 4327– 4329.
- 19. Yu, W.; Mei, Y.; Kang, Y.; Hua, Z.; Jin, Z. Org. Lett. 2004, 6, 3217–3219.
- 20. Hannick, S. M.; Kishi, Y. J. Org. Chem. 1983, 48, 3833–3835.
- 21. RajanBabu, T. V.; Nugent, W. A. J. Am. Chem. Soc. 1994, 116, 986–997.
- 22. Sharpless, K. B.; Umbreit, M. A.; Nieh, M. T.; Flood, T. C. J. Am. Chem. Soc. 1972, 94, 6538–6540.
- 23. Caputo, R.; Mangoni, L.; Neri, O.; Palumbo, G. Tetrahedron Lett. 1981, 22, 3551–3552.
- 24. Kawahara, N.; Nozawa, K.; Nakajima, S.; Udagawa, S.-I.; Kawai, K.-I. Chem. Pharm. Bull. 1988, 36, 398–400.
- 25. Cresp, T. M.; Djura, P.; Sargent, M. V.; Elix, J. A.; Engkaninan, U.; Murphy, D. P. H. Aust. J. Chem. 1975, 28, 2417–2434.